Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 100 KMean σ (C–C) = 0.002 Å R factor = 0.054 wR factor = 0.147 Data-to-parameter ratio = 18.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The title compound, $C_{21}H_{17}N$, was synthesized by the *p*-toluenesulfonic acid-assisted Schiff base reaction between 9-fluorenone and 3,4-dimethylaniline. The crystal structure has been determined at 100 K.

N-9H-Fluoren-9-ylidene-3,4-dimethylaniline

Received 10 June 2004 Accepted 21 June 2004 Online 26 June 2004

Comment

Our interest in imines results from the potential use of chiral amines as auxiliaries in the separation of racemic mixtures of aldehydes and ketones. Chiral amines can form Schiff bases with racemic aldehydes and ketones to form diastereomeric mixtures of imines. These diastereomers can potentially be resolved by either flash chromatography or differential recrystallization.



To test the ease with which imines can be formed, a series of imines were made by reacting various aromatic amines with 9-fluorenone. A previous study has shown that an aromatic amine with an electron-donating group forms the imine with 9-fluorenone readily and in good yield (Glagovich *et al.*, 2004). The title compound, (I), was made in order to test what effects disubstitution of the amines will have on imine formation.

The title compound crystallized with molecules in general positions. The planes defined by the dimethylphenyl and the fluorenylidene moieties are nearly at right angles to each other [89.61 (3)°]. An *ORTEP*-3 (Farrugia, 1997) diagram of (I) is shown in Fig. 1.

Experimental

In a 100 ml round-bottomed flask equipped with a Hickman still and reflux condenser, 2.5 g (14 mmol) 9-fluorenone, 2.04 g (16.8 mmol) 3,4-dimethylaniline, 0.014 g (74 mmol) *p*-toluenesulfonic acid, and 50 ml toluene were combined. The resulting mixture was heated under reflux for 24 h. After this time, the orange solution was concentrated under reduced pressure to produce an orange solid. The solid was purified by flash chromatography (SiO₂, 80% hexanes–ethyl acetate) which produced 3.36 g of (I) as an orange solid (87%). R_F 0.59 (SiO₂, 80% hexanes–ethyl acetate); m.p. 355.8 K; IR (CHCl₃) 3040, 3061, 3017, 1649, 1450, 1376, 1304, 790, 730, 650 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 7.92 (*d*, 1H, *J* = 7.4 Hz), 7.60 (*d*, 2H, *J* = 7.4 Hz), 7.46 (*t*, 1H, *J* = 7.4 Hz), 7.35 (*d*, 1H, *J* = 7.8 Hz), 7.32 (*d*, 1H, *J* = 7.5 Hz), 7.16 (*d*, 1H, *J* = 7.8 Hz), 6.95 (*t*, 1H, *J* = 7.5 Hz), 6.81 (*s*, 1H),

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6.73 (*t*, 1H, *J* = 7.8 Hz), 2.33 (*s*, 3H), 2.29 (*s*, 3H); ¹³C NMR (300 MHz, CDCl₃): δ 162.65, 149.42, 143.73, 141.75, 137.62, 137.48, 132.05, 131.64, 131.55, 131.26, 130.30, 128.33, 127.59, 127.09, 123.21, 120.27, 120.09, 119.50, 115.50, 19.85, 19.29; UV/Vis (CH₂Cl₂; λ_{max} , logε) 404 nm, 2,685; MS calculated for C₂₁H₁₇N: M⁺: 283, measured: 283. Analysis calculated for C₂₁H₁₇N: C 89.01, H 6.05, N 4.94%; found: C 89.26, H 6.18, N 4.81.

Z = 2

 $D_x = 1.251 \text{ Mg m}^{-3}$ Mo *K* α radiation

reflections $\theta = 2.3-28.4^{\circ}$ $\mu = 0.07 \text{ mm}^{-1}$

T = 100 (2) KParallelepiped, orange

Cell parameters from 6350

 $0.40\,\times\,0.34\,\times\,0.10$ mm

Crystal data

C21H17N	
$M_r = 283.36$	
Triclinic, P1	
a = 8.476 (3) Å	
b = 8.870(3) Å	
c = 10.376 (4) Å	
$\alpha = 78.886~(6)^{\circ}$	
$\beta = 79.375 \ (6)^{\circ}$	
$\gamma = 89.483~(6)^{\circ}$	
$V = 752.1 (5) \text{ Å}^3$	
Data collection	

Bruker SMART APEX
diffractometer3722 independent reflections
3472 reflections with $I > 2\sigma(I)$ ω scans $R_{int} = 0.035$ Δ bsorption correction: multi-scan
(SADABS; Sheldrick, 2003) $h = -11 \rightarrow 11$ $T_{min} = 0.967, T_{max} = 0.995$ $k = -11 \rightarrow 11$ 7401 measured reflections $l = -13 \rightarrow 13$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0842P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.054$	+ 0.226P]
$wR(F^2) = 0.147$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.03	$(\Delta/\sigma)_{\rm max} < 0.001$
3722 reflections	$\Delta \rho_{\rm max} = 0.49 \ {\rm e} \ {\rm \AA}^{-3}$
201 parameters	$\Delta \rho_{\rm min} = -0.28 \text{ e} \text{ Å}^{-3}$
H-atom parameters constrained	

All H atoms were placed in calculated positions, with C–H distances of 0.95 and 0.98 Å (for aromatic and methyl H atoms, respectively), and were included in the refinement in the riding-model approximation, with $U_{iso}(H) = 1.2U_{eq}$ (carrier atom).

Data collection: *SMART* (Bruker, 1997–1999); cell refinement: *SMART*; data reduction: *SAINT-Plus* (Bruker, 1997–1999); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997);



Figure 1

A view of (I) (Farrugia, 1997). Displacement ellipsoids are drawn at the 50% probability level. H atoms have been omitted for clarity.

molecular graphics: *ORTEP*-3 (Farrugia, 1997); software used to prepare material for publication: *SHELXL*97.

NG and ER acknowledge AAUP–CSU Research Grants and CCSU Faculty Research Grants for primary support of this research. MZ and JU were supported by NSF grant 0111511, and the diffractometer was funded by NSF grant 0087210, by Ohio Board of Regents grant CAP-491, and by YSU. The authors thank Dr Martha Morton of the University of Connecticut for collecting the NMR spectra.

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